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### Spectroscopic ellipsometry study and variability of optical parameters for blend polymer(PVA:POT)thin films

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#### ABSTRACT

The third-order nonlinear susceptibility of blend poly (O-Toluidine-Vinyl Alcohol) were studied as a function of photon energy. The measurement of refractive index (n) and extinction coefficient (k) showed variation in their amount with incident light. In addition, the some optical parameters of the prepared blended polymers were calculated. The optical and electrical conductivity have been measured as a function of photon energy as well as.

#### Introduction

Thin film fabrication technology is a well-founded materials technology. The request of the desired materials to keep pace with the evolution at twenty-first century, like nanostructured materials and/or handmade super lattices; it is as yet developing on a daily basis. Thin film material technology is both an old and a recent time considered as an important key for material technology studies. Thin film materials and deposition processes with unique properties have been explained in several articles and books. However, there are different technique methods available for thin films fabrication on a single crystal substrate, such as thermal evaporation, spin coating, chemical decomposition, spray pyrolysis and the evaporation of fundamental materials through the radiation of energetic or photons [1, 2]. The studies of the organic conducting polymers started in the last few years for employ in several electronic applications. Especially, the electronic and electrical properties of poly aniline and poly (O-Toluidine) were thoughtful massively [3, 4]. Increasing usage of conducting polymers has quickened studies on polymer applications in recent years. Conducting polymers with long  $\pi$ -conjugated structures have unique properties, such as flexibility, thermal and electrical stability, ease of preparation and immovability [5]. The study of the optical properties for thin films considered as a used tool to understand and explain the behavior of the photonic device applications [6]. In this study of polyvinyl alcohol

(PVA) and blend polymers (PVA:POT) are investigated by using UV-visible spectrophotometer (Cary 50 scan VARIAN) and then the optical and electrical conductivity of polymers are calculated. While the optical constants of polymers, such as refraction index (n), extinction coefficient (k) and third-order nonlinear susceptibility determined by using spectroscopic ellipsometry.

#### Optical studies

The optical constants of prepared polymers (Refractive index (n) and extinction coefficient (k)) are characterized by using Ellipsometry spectroscopy through performing theoretical fitting of the measured ellipsometry data. The Cauchy model has been used to extract the parameters of the polymers in this work. The ration of light velocity in vacuum (C) to it's velocity inside the material (v) known as refractive index. Depending on the reflectance (R) and extinction coefficient (k) the value of refractive index (n) was determined as following [8,9]:

$$n = \left[ \left( \frac{4R}{(R-1)^2} \right) - k^2 \right]^{1/2} - \frac{R+1}{R-1} \dots(1)$$

Where, the reflectance expressed by the following relation:

$$R = \frac{(n-1)^2 + k^2}{(n+1)^2 + k^2} \dots(2)$$

In general, The extinction coefficient refers to several different measures of the light absorption in a medium. In physics, the extinction coefficient is the imaginary part of the complex index of refractive related to light absorption as well. The imaginary part

of complex refractive index (N) which depends on the material type, crystal structure, crystal defects, stress in crystal, and extinction coefficient can be given by the following relations [10-11]:

$$N = n - ik \dots(3)$$

$$k = \alpha\lambda/4\pi \dots(4)$$

The reaction between incident light and charges of the material will be happen as result of light energy which absorbed by the material. This process leads to polarize the charges of the material. The polarization usually described by complex dielectric constant ( $\epsilon$ ) [12].

$$\epsilon = \epsilon_1 - i\epsilon_2 \dots(5)$$

Where,  $\epsilon_1$ ,  $\epsilon_2$  are the real and imaginary part of the complex dielectric constant respectively.

The dielectric constant in its two parts can be determined by using the following expressions [13,14]:

$$\epsilon = (N)^2 \dots(6)$$

$$\epsilon_1 = (n^2 - k^2) \dots(7)$$

$$\epsilon_2 = (2nk) \dots(8)$$

The optical and electrical conductivity is a property of material to conduct the electrical current through it as result of incident light and it depends on the intensity of optical radiation [15]. The optical and electrical conductivity ( $\sigma_{opt.}$ ) ( $\sigma_{ele.}$ ) given by the following relations respectively:

$$\sigma_{opt.} = \alpha n c/4\pi \dots(9)$$

$$\sigma_{ele.} = 2\lambda\sigma_{opt.}/\alpha \dots(10)$$

Where c is the velocity of light.

The refractive index dispersion can be analysed by the single-oscillator model [16].

$$(n^2 - 1)^{-1} = \frac{E_o}{E_d} - \frac{1}{E_o E_d} E^2 \dots(11)$$

Where  $E_o$  is the oscillator energy, E the photon energy and  $E_d$  is the dispersion energy, which measures the average strength of the interband optical transition. The moments of optical dispersion spectra,  $M_{-1}$  and  $M_{-3}$ , can be evaluated using the relations [17, 18]:

$$E_o^2 = \frac{M_{-1}}{M_{-3}} \dots(12)$$

$$E_d^2 = \frac{M_{-1}^3}{M_{-3}} \dots(13)$$

The third order nonlinear susceptibility ( $\chi^3$ ) has been calculated from the following equation [19]:

$$\chi^{(3)} = A \left\{ E_o \frac{E_d}{4\pi(E_o^2 - (h\nu)^2)} \right\}^4 = A / (4\pi)^4 (n - 1)^4 \dots(14)$$

Where A is constants equal to  $1.7 \times 10^{-10}$ .

### Experimental

The chemical polymerization method was used to synthesis poly (o-toluidine). The o-toluidine monomer (purchased from Fisher scientific) with molarity concentration of (0.27M) was dissolved in acidic medium of 0.25M hydrochloric acid. The temperature of acidic medium was controlled to be at range of (0-5°C) through ice medium. The ammonium per sulphate (purchased from sigma-Aldrich) with molarity concentration of 0.54M was wisely dropped to the continuously stirred o-toluidine solution as oxidizing agent. The poly (o-toluidine) as greenish-black precipitate was filtered, washed with distilled water , methanol and acetone respectively, and then dried under 60°C. The polymer powder grinded and sieved was then blended with PVA (purchased from sigma-Aldrich) (1:1, 1:2, 2:1 V/V) solution in formic acid and spun into Nano thin films to obtain information about optical parameters.[7].

### Results and discussion

Fig. 1 shows behavior of the refractive (n) and extinction coefficient (k) for prepared blend polymers. The refractive index (n) of pure PVA is decreased at a high wavelength while n of blend polymers at different volume ratios (1:1, 1:2, and 2:1 PVA:POT) becomes high at large values of wavelength. The extinction coefficient (k) of pure PVA is decreased at low wavelength values but in case of blend polymer increased. The values of both n and k increased at high and low wavelength relating to the present of POT that also lead to increase in the extinction index value compared with pure PVA. Thus, the large value of refractive index (n) leading to high absorption of incident light by compound of polymers and the changes in value of refractive index with respect to variation of wavelength according to equation (4) [11].

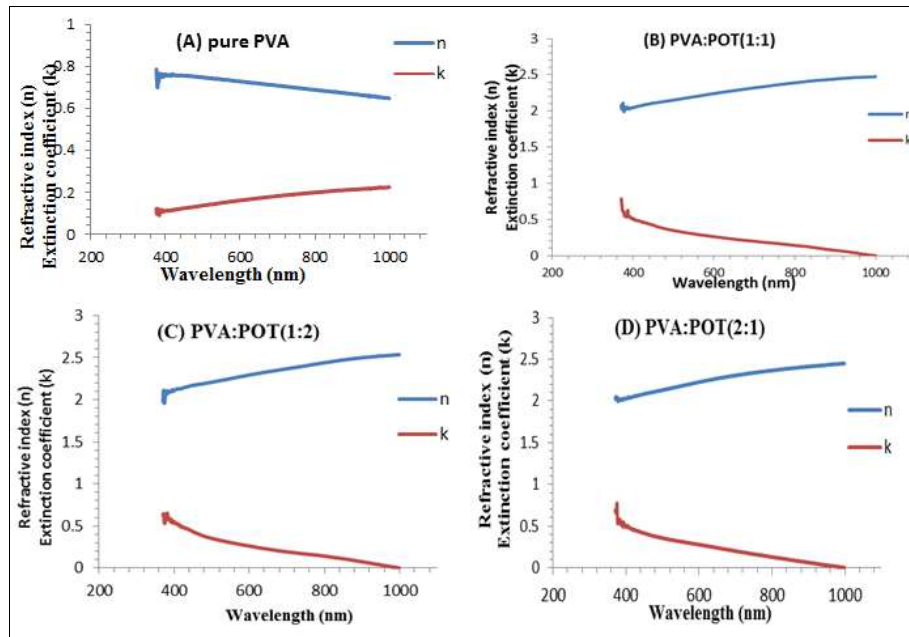


Fig.(1): relation between refractive index (n) / extinction coefficient (k) and wavelength of blend polymers PVA:POT.

The equations (7) and (8) are used to calculate the real and imaginary complex dielectric constant respectively. Real and imaginary part of the complex dielectric constant for the pure PVA is illustrated in fig. 2. The behaviour of real and imaginary dielectric constant approximately same the refractive index because of the value of extinction coefficient is low compare with refractive index value. The real and imaginary part of complex dielectric constant as shown in fig. 3 and 4, are decreased and increased with increasing photon energy respectively.

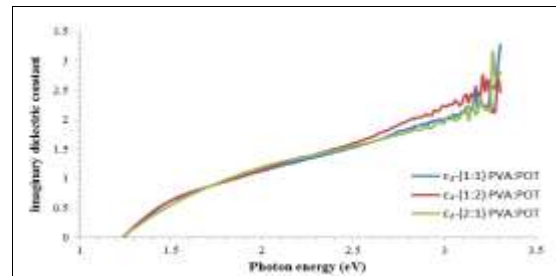


Fig.(4): shows imaginary part of complex dielectric constant as function of photon energy for blend polymers (PVA:POT).

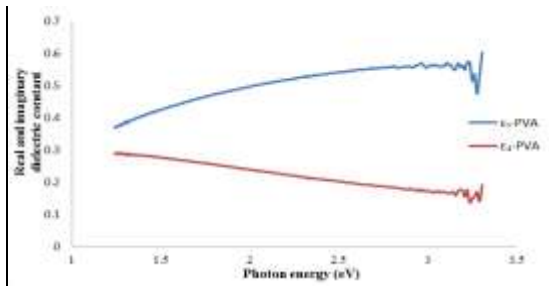


Fig.(2): shows real and imaginary part of complex dielectric constant as function of photon energy for pure PVA.

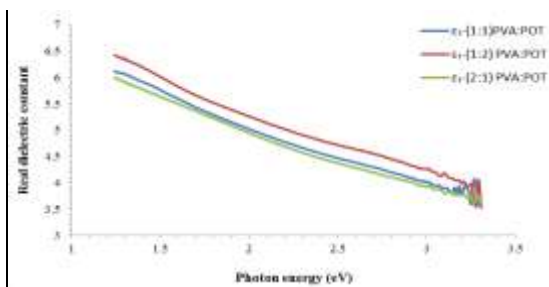


Fig.(3): shows real part of complex dielectric constant as function of photon energy for blend polymers (PVA:POT)

Fig. 5 shows variation of optical conductivity calculated according to equation (9) with photon energy in which the maximum conductivity was recorded at 1.56 eV to be  $(1.75, 2.12 \text{ and } 2.5 \times 10^{-14})$  for (2:1, 1:1 and 1:2 PVA:POT) respectively. The obtained increase in conductivity attributed to present POT as a larger ratio of POT that leads to forming additional charges in polymers. The electrical conductivity decreased with increased photon energy as illustrated in fig. 6. The optical and electrical conductivity of the polymer blend varies with increased amount of POT. Another interesting result is that the optical conductivity exhibit two peaks where the electrical conductivity without peaks. The origin of these two peaks may be refers to optical interband transitions [20].

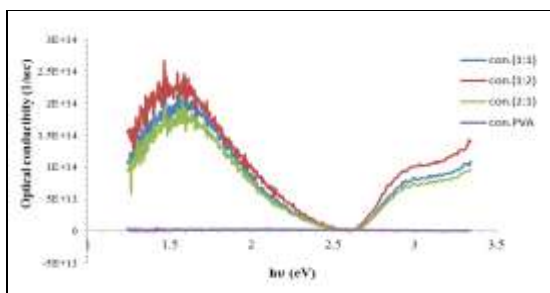


Fig.(5): optical conductivity versus photon energy of blend polymers (PVA:POT).

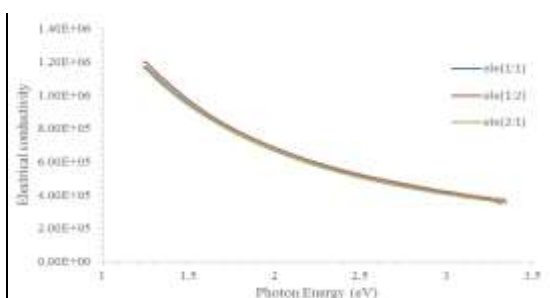


Fig.(6): Electrical conductivity versus photon energy of blend polymers (PVA:POT).

Fig. 7 show the relation between  $(n^2 - 1)^{-1}$  and  $E^2$  for the investigated composition. The values of  $E_d$  and  $E_o$  can be determined from the slope and the intercept. The calculated values of dispersion energy  $E_d$ , the oscillator energy  $E_o$  are found to be as listed in the table. 1.

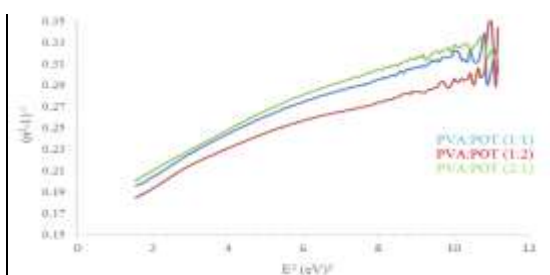


Fig. (7): The relation between  $(n^2 - 1)^{-1}$  and  $E^2$  of blend polymer (PVA:POT).

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Table 1: Optical parameters of blend polymer (PVA:POT).

material	$E_o$	$E_d$	$M_{-1}$	$M_{-3}$
PVA:POT (1:1)	3.53	19.63	5.56	0.446
PVA:POT (1:2)	3.59	20.98	5.84	0.453
PVA:POT (2:1)	3.50	19.11	5.45	0.443

The fig. 8 shows relation between the third order nonlinear susceptibility as a function of photon energy. The third order nonlinear susceptibility was calculated according to equation 14, and increased with increasing the ratio of POT in blended polymers which gave us a pointer, the POT enhanced the value of third order nonlinear susceptibility. Moreover, its value decreases with increase the incident photon energy.

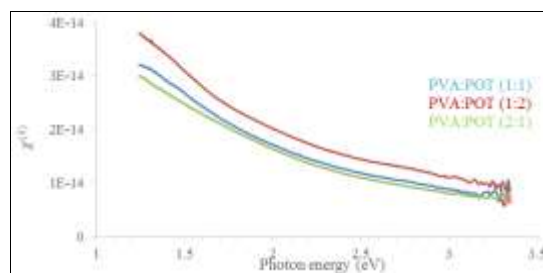


Fig. (8): The relation between the third order nonlinear susceptibility and  $E$  of blend polymer (PVA:POT).

## Conclusion

In this work, the refractive index ( $n$ ) and extinction coefficient ( $k$ ) were characterized by using Ellipsometry spectroscopy through performing theoretical fitting of the measured ellipsometry data. The optical and electrical conductivity of blend polymer (PVA:POT) have been studied. The maximum conductivity was recorded at 1.56 eV to be (1.75, 2.12 and 2.5 for (2:1, 1:1 and 1:2 PVA:POT) respectively. The oscillator energy ( $E_o$ ) and dispersion energy ( $E_d$ ) were measured according to single-oscillator model. The moments of optical dispersion spectra,  $M_{-1}$  and  $M_{-3}$  evaluated as well as. The third order nonlinear susceptibility ( $\chi^3$ ) has been calculated in this work and studied as a function of the photon energy.

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## دراسة المتغيرات البصرية من خلال تحليل اطياف الاليسومتري لأغشية البوليمرات الممزوجة البولي

### اورثو تولدين/البولي فنيل الكحول

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### الملخص

تمت في هذه الدراسة حساب بعض الثوابت البصرية للبوليمرات الممزوجة (البولي تولدين/ البولي فنيل الكحول) من خلال تحليل اطياف الاليسومتري، وظهرت الدراسة تباينا في مقدار معامل الانكسار ومعامل الخمود كدالة لطاقة الفوتون، وكذلك تم حساب كلا من التوصيلية البصرية والتوصيلية الكهربائية والتأثرية اللاخطية من الرتبة الثالثة كدالة لطاقة الفوتون. ايضا في هذه الدراسة تم حساب قيم بعض الثوابت البصرية (طاقة الاهتزاز، طاقة التشتت، وزخم اطياف التشتت البصري).